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1971). The structural elucidation via NMR spectra has already been carried out (Achari, Pal & Pakrashi, 1975). The present X-ray structure study was carried out in order to confirm these findings. Isoalangidiol is the 3α -epimer of 18α -B':A'-neogammacerane-3,18-diol. It represents the first example of a naturally occurring D:Ecis neohopane derivative and belongs to the rare group of pentacyclic triterpenes with a free hydroxyl group at the ring juncture.



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Isoalangidiol Monoacetate, a Triterpene Alcohol

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Abstract

Isoalangidiol $(3\alpha, 18\alpha-B':A'$ -neogammacerane-3,18-diol) has been extracted as a natural product from the leaves of *Alangium lamarckii Thw.* (Alangiaceae). The crystal structure study of its monoacetate, $C_{32}H_{54}O_3$, was undertaken in order to ascertain the conformation of the compound. The four six-membered rings are in chair conformations. The five-membered ring is distorted from a plane and is twisted, with a C19—C18— C17—C21 torsion angle of -43.0 (4)°. The molecular parameters of the compound are all within normal limits.

Comment

Isoalangidiol has been obtained from the petroleum ether extract of *Alangium lamarckii Thw*. (Alangiaceae) and purified as its monoacetate, (I), prepared by heating with acetic anhydride/pyridine at 373 K, by column chromatography over silica gel (Pakrashi & Achari,

© 1997 International Union of Crystallography Printed in Great Britain – all rights reserved The title compound consists of four *trans*-fused sixmembered alicyclic rings (A, B, C and D) and a fivemembered ring E cis-fused to ring D. Rings A, B, C and D are in chair conformations, as shown by the ring-puckering parameters given in Table 2 (Cremer & Pople, 1975). The acetoxy function, adjacent to a methylene group, is axial. There are altogether eight methyl groups in the structure. The isopropyl group is attached to ring E. The hydroxy group is attached to C18 in this neohopane derivative. The five-membered ring has a twisted conformation, with a C19—C18— C17—C21 torsion angle of $-43.0(4)^{\circ}$.

There is an intermolecular O—H···O hydrogen bond between the O18 and O32 atoms, joining the molecules into infinite chains along the z axis [O18···O32ⁱ 2.961 (5) Å and O18—H18···O32ⁱ 124.1 (4)°; symmetry code: (i) $-x - \frac{1}{2}, -y, z - \frac{1}{2}$].



Fig. 1. The structure of (I) showing 30% probability displacement ellipsoids and the atom-numbering scheme.

Experimental

Isoalangidiol monoacetate, (I), was crystallized from a solution of benzene and ethanol.

Crystal data

$C_{32}H_{54}O_3$	Mo $K\alpha$ radiation
$M_r = 486.75$	$\lambda = 0.71073 \text{ Å}$

$C_{32}H_{54}O_3$

Orthorhombic $P2_12_12_1$ a = 12.201(2) Å	Cell parameters from 25 reflections $\theta = 10-15^{\circ}$	C27C14C8 C13C14C8 C28C17C16 C28C17C21	111.5 (3) 108.2 (3) 108.4 (3) 111.0 (4)	O32—C32—O31 O32—C32—C33 O31—C32—C33	125.2 (5) 124.2 (5) 110.5 (5)
b = 12.8625 (14) A c = 18.571 (2) Å $V = 2914.4 (7) \text{ Å}^3$ Z = 4 $D_x = 1.109 \text{ Mg m}^{-3}$ D_m not measured	$\mu = 0.069 \text{ mm}^{-1}$ T = 293 (2) K Tablet $0.28 \times 0.24 \times 0.10 \text{ mm}$ Colourless	$\begin{array}{c} C10-C1-C2-C3\\ C1-C2-C3-O31\\ C1-C2-C3-C4\\ C2-C3-C4-C24\\ O31-C3-C4-C24\\ O31-C3-C4-C5\\ C2-C3-C4-C23\\ C3-C4-C5-C6\\ C24-C4-C5-C10\\ \end{array}$	56.3 (5) 66.5 (5) -53.2 (5) -77.3 (5) -71.5 (5) 166.7 (4) 177.5 (4) 70.9 (5)	C9-C11-C12-C13 C11-C12-C13-C18 C12-C13-C14-C15 C18-C13-C14-C8 C9-C8-C14-C15 C7-C8-C14-C13 C9-C8-C14-C13 C9-C8-C14-C13 C8-C14-C15-C16	56.3 (5) 169.1 (3) - 178.8 (3) - 170.2 (3) - 179.1 (3) - 178.7 (3) - 59.5 (4) 171.6 (3)
Data collection		C10-C5-C6-C7	62.8 (5)	C14-C15-C16-C17	-58.9(5)
Enraf-Nonius CAD-4 diffractometer ω -2 θ scans Absorption correction: none	$\theta_{max} = 22.96^{\circ}$ $h = 0 \rightarrow 13$ $k = 0 \rightarrow 14$ $l = 0 \rightarrow 20$	C4—C5—C6—C7 C5—C6—C7—C8 C6—C7—C8—C14 C7—C8—C9—C11 C14—C8—C9—C11	-162.7 (4) -57.3 (5) 167.2 (3) 177.6 (3) 57.2 (4) 167.6 (3)	C15-C16-C17-C21 C14-C13-C18-C19 C14-C13-C18-C17 C21-C17-C18-C13 C16-C17-C18-C19 C14-C17-C18-C19	59.5 (5) 161.6 (3) 48.2 (4) 75.3 (4) 166.1 (3) 43.0 (4)
2315 measured reflections	3 standard reflections	$C_2 = C_1 = C_1 (0 = -C_2)$	-107.6(3) -1736(4)	$C_{21} - C_{13} - C_{13} - C_{19}$	-43.0(4)
2315 independent reflections	every 100 reflections	C4C5C10C9	168.0 (3)	C19-C20-C21-C22	-145.3 (4)
1853 reflections with $I > 2\sigma(I)$	intensity decay: 1%	C8-C9-C10-C1 C11-C9-C10-C5 C10-C9-C11-C12 C8-C9-C11-C12	$ \begin{array}{r} 168.4 (3) \\ -173.8 (3) \\ 169.6 (3) \\ -55.1 (5) \end{array} $	C19C20C21C17 C16C17C21C20 C2C3O31C32 C4C3O31C32	-15.9(5) 154.6(4) 82.9(5) -153.5(4)

Refinement

-2

$(\Delta/\sigma)_{\rm max} = 0.063$
$\Delta \rho_{\rm max} = 0.219 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.206 \ {\rm e} \ {\rm \AA}^{-3}$
Extinction correction: none
Scattering factors from
International Tables for
Crystallography (Vol. C)
Absolute configuration:
Flack (1983)
Flack parameter = $0(3)$

Table 1. Selected geometric parameters (Å, °)

C1-C10	1,540(6)	C14-C15	1 546 (5)
C3-031	1 465 (5)	C16-C17	1 537 (6)
C3C4	1.535 (7)	C17-C28	1 527 (6)
C4-C24	1524(7)	$C_{17} - C_{21}$	1.562 (6)
C4	1.562 (6)	C17-C18	1.502 (6)
C4-C23	1.564 (8)	C18-018	1.371 (0)
C5-C10	1.551 (6)	C18-C19	1.536 (6)
C7-C8	1.535(6)	C19-C20	1.530 (0)
$C8 - C^{26}$	1.550(5)	C_{20}	1.542 (6)
C8-C9	1.550 (5)	C_{21} C_{21}	1.543 (6)
C8-C14	1.594 (5)	C^{22} C^{30}	1.540 (6)
C9-C10	1.572 (6)	C^{22} C^{29}	1.575 (6)
C10-C25	1.572 (6)	031 - 032	1 328 (6)
C13-C18	1.539(6)	$C_{32} = 0_{32}$	1.202 (6)
C13-C14	1.562 (5)	C32-C33	1 507 (7)
C14—C27	1.544 (5)	0.2 0.0	
011 01 01	100 5 (4)	014 017 011	
031 - 03 - 02	108.5 (4)	C16C17C21	116.0 (3)
031 - C3 - C4	107.5 (4)	C28—C17—C18	110.2 (4)
C2—C3—C4	113.9 (4)	C16C17C18	109.5 (3)
C24—C4—C5	115.3 (4)	C21—C17—C18	101.6 (3)
C3-C4-C5	109.2 (4)	O18—C18—C13	107.3 (3)
C24—C4—C23	107.5 (4)	O18—C18—C19	112.0 (3)
C3—C4—C23	108.1 (4)	C13-C18-C19	110.2 (3)
C7—C8—C26	107.3 (3)	O18—C18—C17	112.5 (3)
С7—С8—С9	108.6 (3)	C13—C18—C17	112.6 (3)
С26—С8—С9	111.6(3)	C19C18C17	102.2 (3)
C26—C8—C14	109.9 (3)	C20-C19-C18	105.9 (4)
C1—C10—C25	107.8(3)	C19—C20—C21	107.7 (4)
C1-C10-C5	107.2 (3)	C22—C21—C20	111.9 (4)
C25—C10—C5	115.0 (4)	C22C21C17	118.8 (4)
C25—C10—C9	112.7 (4)	C20—C21—C17	104.0 (3)
C5—C10—C9	106.1 (3)	C30C22C29	108.5 (3)
C27—C14—C15	106.2 (3)	C30-C22-C21	115.0 (4)
C27—C14—C13	110.8 (3)	C29—C22—C21	109.8 (4)
C15—C14—C13	108.8 (3)	C32-O31-C3	117.2 (4)

Table	2.	Ring	puck	ering	parar	neters	for	four	rings	of	the
				title	com	oound					

	q_2 (Å)	q_3 (Å)	<i>Q</i> τ (Å)	θ(°)
Ring A	0.038 (5)	-0.534(5)	0.535 (5)	175.9 (6)
Ring B	0.094 (5)	-0.564(4)	0.572 (4)	170.6 (4)
Ring C	0.056 (4)	-0.606(4)	0.609 (4)	174.7 (4)
Ring D	0.057 (4)	0.530(4)	0.533 (4)	6.1 (4)

Ring-puckering parameters were calculated using PARST (Nardelli, 1983). All 54 H atoms were introduced at calculated positions and refined isotropically, riding on their parent atoms, using the AFIX function in SHELXL93 (Sheldrick, 1993).

Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: CAD-4 Software. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93. Molecular graphics: ORTEP (Johnson, 1965). Software used to prepare material for publication: SHELXL93.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: DE1038). Services for accessing these data are described at the back of the journal.

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The 4:2 Diels–Alder Adduct of 1,3-Cyclopentadiene with 1,4-Naphthoquinone

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Abstract

The title compound, 1,4-methano-1,4,4a,9a-tetrahydroanthracene-9,10-dione, $C_{15}H_{12}O_2$, is a Diels-Alder adduct resulting from a [4+2] cycloaddition of 1,3cyclopentadiene with *p*-naphthoquinone.

Comment

The structure analysis of the title compound is part of an ongoing program of obtaining cage compounds by photochemical [2+2] cycloaddition (Maruyama, Terada & Yamamoto, 1981; Kaftory & Weisz, 1984). One aim is to obtain the norbornediene derivatives from the title compound (Diels & Alder, 1929). The present structural analysis has been carried out in order to establish more details of the molecular geometry of the title compound, (I), thus providing a basis of comparison for our further studies.



The unit cell contains two independent molecules, related by a pseudo-c glide perpendicular to the a axis, in the asymmetric unit, the average deviation from cglide symmetry being 0.02 (2) Å between the two molecules. The C and O atoms of each dihydronaphthoquinone ring are coplanar (Pizzotti, Cenini, Ugo & Demartin, 1991). The r.m.s. deviations of fitted atoms for these planes are 0.025 and 0.057 Å, the maximum deviations from these planes are 0.04 (3) Å for C6 and 0.164 (3) Å for C1A. The two five-membered rings adopt envelope conformations with C15 and C15A at the flaps. The dihedral angles between the envelope planes through C11–C14 and C11A–C14A, respectively, and the dihydronaphthoquinone moieties are 55.5(1) and 49.7(2)° for each independent molecule. The dihedral angle is 29.8(2)° between the two envelope planes and 57.7(1)° between the two dihydronaphthoquinone moieties in the asymmetric unit.

The bond lengths and angles are quite normal and comparable with corresponding values observed in related molecules (Kerr, 1987; Beddoes, Gorman & Mc-Neeney, 1993).



Fig. 1. A perspective view of the molecular structure of the title compound with the atom-numbering scheme. The displacement ellipsoids are drawn at the 50% probability level.

Experimental

p-Naphthoquinone (31.63 g, 0.20 mol) was suspended in benzene (100 ml). A cyclopentadiene (13.88 g, 0.21 mol) solution in cooled benzene (20 ml) was added to the above suspension. After keeping the mixture overnight, the resulting colourless solid was separated and recrystallized from ethanol.

Crystal data

C₁₅H₁₂O₂ $M_r = 224.25$ Triclinic $P\overline{1}$ a = 9.4077 (12) Å b = 10.1752 (10) Å c = 12.5542 (13) Å $\alpha = 71.698 (9)^{\circ}$ $\beta = 86.980 (9)^{\circ}$ $\gamma = 77.019 (10)^{\circ}$ $V = 1111.6 (2) Å^{3}$ Z = 4 $D_x = 1.340 \text{ Mg m}^{-3}$ $D_m \text{ not measured}$ Mo $K\alpha$ radiation $\lambda = 0.71073$ Å Cell parameters from 25 reflections $\theta = 9-18^{\circ}$ $\mu = 0.088$ mm⁻¹ T = 293 (2) K Prismatic $0.64 \times 0.44 \times 0.24$ mm Colourless